

## HS-014-003201

Seat No.

## Masters in Pharmacy Management (MPM) (Sem. II) (CBCS) Examination

June / July - 2017 Physical Pharmacy

Faculty Code: 014 Subject Code: 003201

Time: 3 Hours [Total Marks: 80

**Instructions**: 1. Answer and tie both the sections separately.

- 2. Figure to the right indicates marks.
- 3. Answer any three (3) questions from each section.
- 4. Que. One (1) and Que. Five (5) are compulsory.
- 5. Draw neat and clean diagrams as required.

## **SECTION - I**

- 1 Answer any seven out of given ten questions :  $7\times2=14$ 
  - a) Explain Raoult's Law.
  - b) Give difference between Flocculated and Deflocculated Suspensions.
  - c) Enlist different identification test for types of emulsion.
  - d) Difference between physical adsorption and chemical adsorption.
  - e) Define: Thixotropy.
  - f) Explain Krafft Point.
  - g) Define chelates and give examples of it.
  - h) Define Surface tension and Interfacial tension.
  - i) What do you mean by ideal solution and real solution?
  - j) What is polymorphism? Enlist its importance with example.
- **2** Answer the following :
  - a) Describe methods for measurement of particle size.
  - b) Define adsorption isotherm. Draw various types of adsorption isotherm and explain in detail.

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3	Answer the following:			
	a)	Enumerates the method for analysis of	7	
		complexation and explain in detail about solubil	ity	
		method.		
	b)	Describe in detail about theory of emulsification.	6	
4	Answer the following:			
	a)	Explain Solubility of Liquid in Liquid.	7	
	b)	Write a note on physical stability of emulsion.	6	
		SECTION - II		
5	Answer any two out of given three questions : 2×7=1			
	a)	Describe various methods of achieving liquefaction of	of gases.	
	b)	Classify the complexes and explain in detail any o	ne.	
	c)	Explain Liquid Crystalline State.		
6	Ans	Answer the following:		
	a)	Describe Newtonian and non Newtonian system.	7	
	b)	How will you determine Rheological Properties?	6	
7	Ans	Answer the following:		
	a)	Write a brief account on Surface active agents.	7	
	b)	Give general principles of Solubility.	6	
8	Answer the following:			
	a)	Discuss the optical properties of colloids.	7	
	b)	Discuss two component systems containing phases	6	
	•	giving proper examples.		